

# Darcy Constant for Multisized Spheres with No Arbitrary Constant

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MacDonald et al. (1991) have presented a model for flow resistance in a bed of multisized spheres. Their derived Darcy permeability  $k$  is a function of the first and second moments of the particle-size distributions

$$k = \frac{1}{180} \frac{\epsilon^3}{(1-\epsilon)^2} \left( \frac{M_2}{M_1} \right)^2 \quad (1)$$

where  $\epsilon$  is the bed porosity and  $M_1$  and  $M_2$  are the first and second moments. The constant (1/180) was used as the best number to reconcile theory and experiment. Reasonable agreement between Eq. 1 and the experiment was found (see MacDonald et al., 1991, Figure 5).

We propose that the Darcy permeability  $k$  for spheres can be derived from first principles and without recourse to arbitrary constants such as (1/180). Happel and Brenner (1973) have provided an expression for  $k$  when flow is around spheres of radius  $R$

$$k = R^2 \left( \frac{2}{9\gamma^3} \right) \left( \frac{3 - \frac{9}{2}\gamma + \frac{9}{2}\gamma^5 - 3\gamma^6}{3 + 2\gamma^5} \right) \quad (2)$$

Here  $R$  is the sphere radius and  $\gamma^3 = (1-\epsilon)$ . We now show how this can be generalized to multisized systems.

For a multisized system, the  $j$ th moment can be written as

$$M_j = \sum_i D_i^j n_i \quad (3)$$

where  $n_i$  is the number of particles with diameter  $D_i$ .

The total area ( $S_T$ ) of a multisized system, given that the number and size of each particle fraction present is known, is given by

$$S_T = \sum_i n_i S_i = \sum_i n_i \pi D_i^2 \quad (4)$$

where  $S_i$  is the area of a single sphere of diameter  $i$ .

In terms of the particle moments for a multisized system, a diameter  $D_m$  can be defined as

$$D_m^2 = \frac{S_T}{\pi n_T} = \frac{\pi \sum_i n_i D_i^2}{\pi \sum_i n_i} = \frac{M_2}{M_0} \quad (5)$$

where  $n_T$  is the total number of particles. For a multisized system, if the weight fraction  $x_i$  of each size is known, the total mass of the system is  $m_T$ , and the particle density is  $\rho_P$ , then

$$x_i m_T = n_i \frac{\pi}{6} D_i^3 \rho_P \quad (6)$$

giving

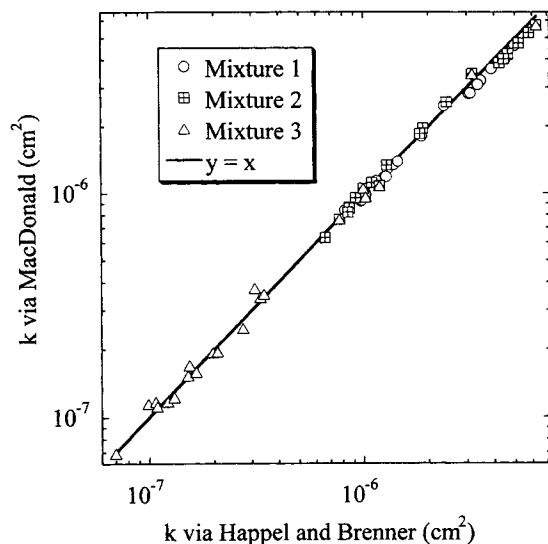
$$n_i = \frac{x_i m_T}{\frac{\pi}{6} D_i^3 \rho_P} \quad (7)$$

Substituting Eq. 7 into Eq. 5 and canceling the terms gives

$$D_m^2 = \frac{\sum_i \frac{x_i}{D_i}}{\sum_i \frac{x_i}{D_i^3}} \quad (8)$$

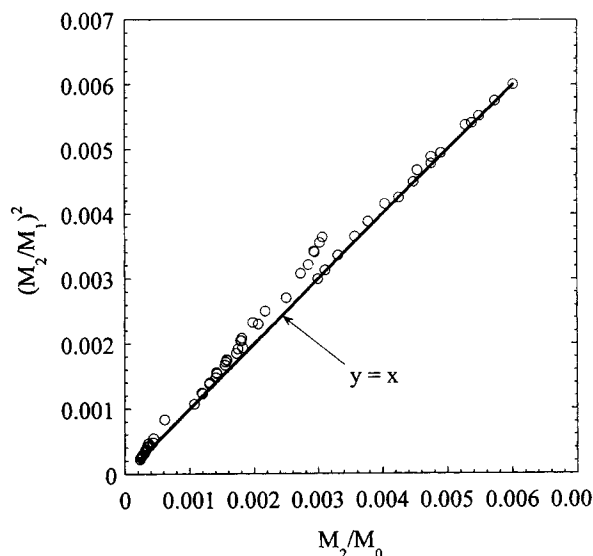
Replacing  $D_m^2$  by  $4R_m^2$ , Eq. 8 can then be substituted into Eq. 2 to give (substituting  $(1-\epsilon)$  for  $\gamma^3$  at the same time)

$$k = \left( \frac{1}{12} \right) \left( \frac{3(1-\epsilon)^{5/3} - 3(1-\epsilon)^{1/3} - 2(1-\epsilon)^2 + 2}{(1-\epsilon)[2(1-\epsilon)^{5/3} + 3]} \right) \times \left( \frac{\sum_i \frac{x_i}{D_i}}{\sum_i \frac{x_i}{D_i^3}} \right) \quad (9)$$



**Figure 1. Comparison of  $k$  derived: MacDonal (1991, Eq. 1) vs. theory of Happel and Brenner (1973, Eq. 9).**

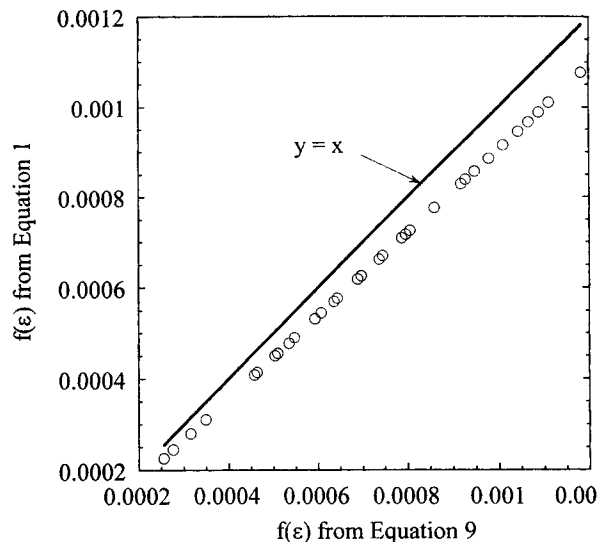
Values for  $\epsilon$ ,  $D_i$  and  $x_i$  were obtained from the three mixtures used by MacDonal.



**Figure 2. Comparison of the square of the mean particle diameter calculated from the particle size distribution moments according to Eq. 1,  $(M_2/M_1)^2$ , and Eq. 5,  $M_2/M_0$ .**

Equation 9 contains no arbitrary constants. Figure 1 shows a comparison between the MacDonal  $k$  values (Eq. 1) and Eq. 9 for the three mixtures used in the MacDonal study. The two results are nearly identical.

The close agreement between Eqs. 1 and 9 for the system in question is not surprising when the relationships are exam-



**Figure 3. Comparison of the porosity term  $f(\epsilon)$  from Eq. 1,  $\left(\frac{1}{180} \frac{\epsilon^3}{(1-\epsilon)^2}\right)$ , and Eq. 9,  $\left(\frac{1}{12}\right)$**

$$\times \left( \frac{3(1-\epsilon)^{5/3} - 3(1-\epsilon)^{1/3} - 2(1-\epsilon)^2 + 2}{(1-\epsilon)[2(1-\epsilon)^{5/3} + 3]} \right).$$

ined more closely. Each equation has a term that is a function of the total porosity, and a term that essentially calculates an effective particle radius based on the particle-size distribution. Figure 2 compares the result of the effective particle radius for the two equations. For the systems studied, the two methods are nearly identical (average error of 8%) for almost all of the mixtures.

The porosity terms in the two equations are compared in Figure 3. Again, agreement between the relatively simple relation of Eq. 1 and the more complicated relation of Eq. 9 is good. In fact, agreement would be almost perfect if the constant in Eq. 1 were  $(1/162)$  instead of  $(1/180)$ .

Equation 9 thus provides a direct and unambiguous method for determining the resistance to flow of a mixture of spheres. No arbitrary constant is needed.

## Acknowledgments

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## Literature Cited

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